

In each box, write the formula of the ionic compound consisting of the positive ion to the left of the box and the negative ion above the box. Check your answers with the interactives on the lesson page.

	Cl^-	S^{2+}	F^-	N^{3-}	O^{2-}	P^{3-}
Mg^{2+}	MgCl_2					
CS^+						
CR^{3+}						
NA^+						
ZN^{2+}						
AL^{3+}						

TERNARY COMPOUNDS

	OH^-	SO_4^{2-}	NO_3^-	CO_3^{2-}	PO_4^{3-}
MN^{2+}					
K^+					
AL^{3+}					
NH_4^+					
CA^{2+}					
FE^{3+}					

FROM

[HTTP://CMS.GAVIRTUALSCHOOL.ORG/SHARED/SCIENCE/CHEMISTRY_SHARED_17/CHEMISTRY_FORMULAS_EQUATIONS_SHARED/CHEMISTRY_FORMULAS_EQUATIONS_SHARED3.HTML](http://cms.gavirtualschool.org/shared/science/chemistry_shared_17/chemistry_formulas_equations_shared/chemistry_formulas_equations_shared3.html)