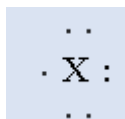
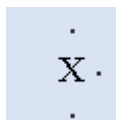


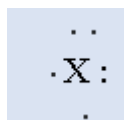
Boron



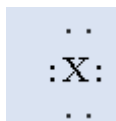
Iodine



Krypton



Magnesium



Sulfur

Match the columns.

Na and F

$3.5 - 1.0 = 2.5$  Ionic

N and O

$3.5 - 3.0 = 0.5$  Polar covalent

I and I

$2.8 - 2.5 = 0.3$  nonpolar covalent

Fe and Cl

$4.0 - 0.9 = 3.1$  Ionic

Br and I

$2.5 - 2.5 = 0$  Nonpolar covalent

Ca and O

$3.0 - 1.8 = 1.2$  Polar covalent

Draw Lewis Structures for the following molecules and polyatomic ions. Check your answer with the interactive below.

